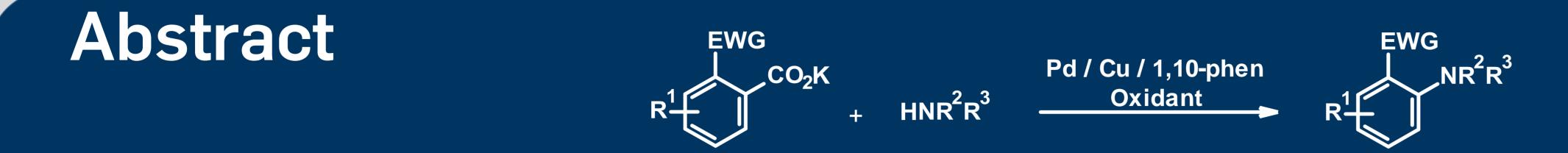


Decarboxylative *ipso*-Amination of Benzoic Acids

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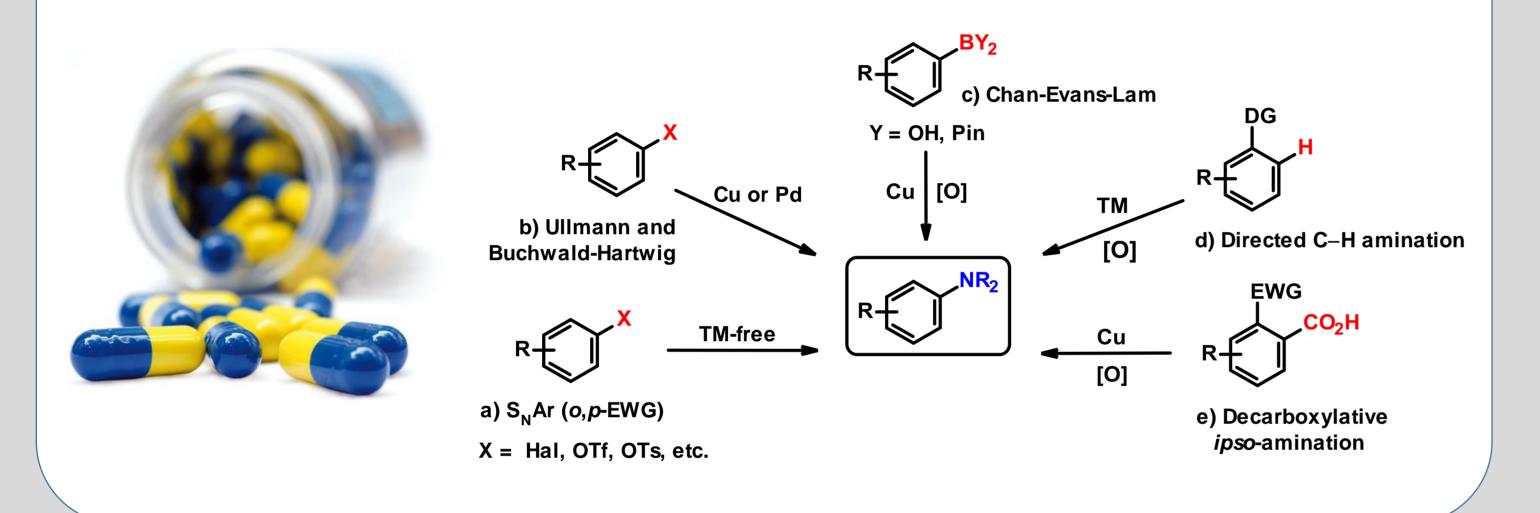


Using a bimetallic Pd/Cu-system with 1,10-phenanthroline as ligand and air or NMO as oxidants, electron-poor benzoic acids undergo oxidative decarboxylative coupling with free amines. This operationally simple aniline synthesis is widely applicable with regard to the amine and gives good yields even on multi-gram scale. Multisubstituted arenes are accessed via orthogonal C–C, C–Cl and C–N functionalizations. Preliminary mechanistic investigations suggest that arylamine formation occurs via reductive elimination from Pd^{IV} intermediates.

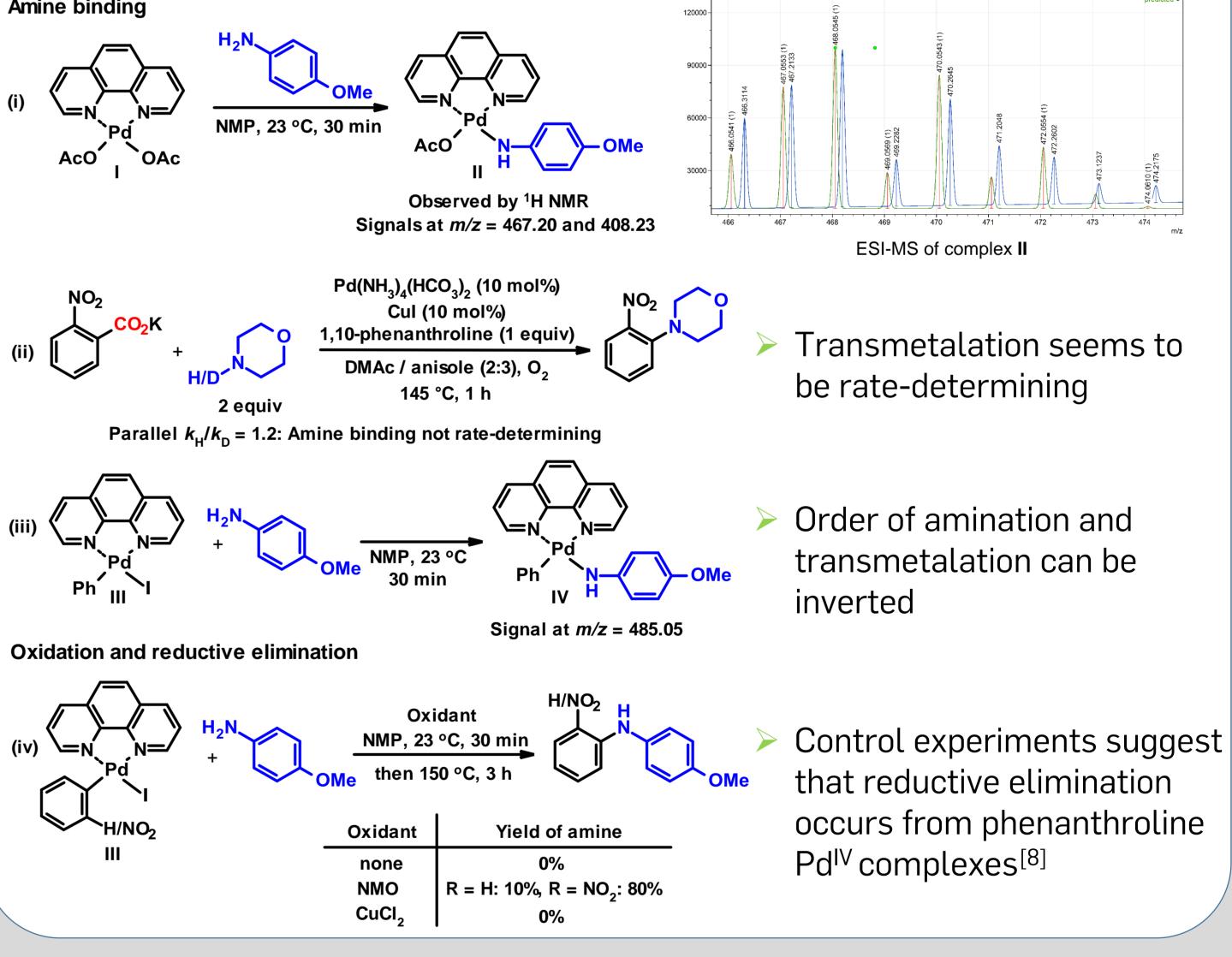
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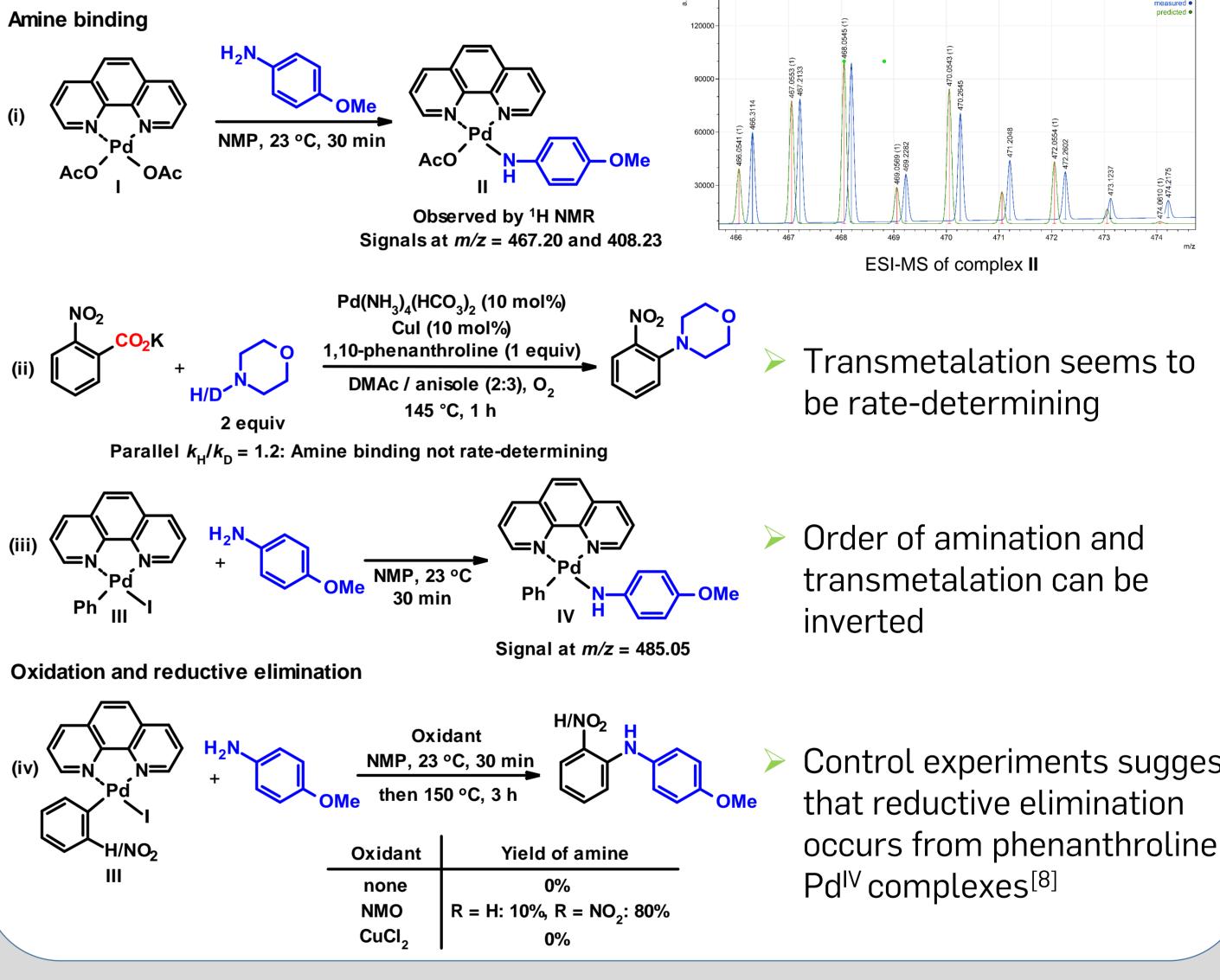
N-Alkylanilines via Decarboxylative Couplings

Substituted anilines^[1,2] are ubiquitous in pharmaceuticals^[3] and material science.^[4] The forcing conditions of aniline synthesis^[5-6] imply the possibility of many side reactions of the basic amines, such as thermal condensation of oxidative ammonium benzoates, oxidative dimerization of anilines, decomposition of aliphatic amines and protodecarboxylation.

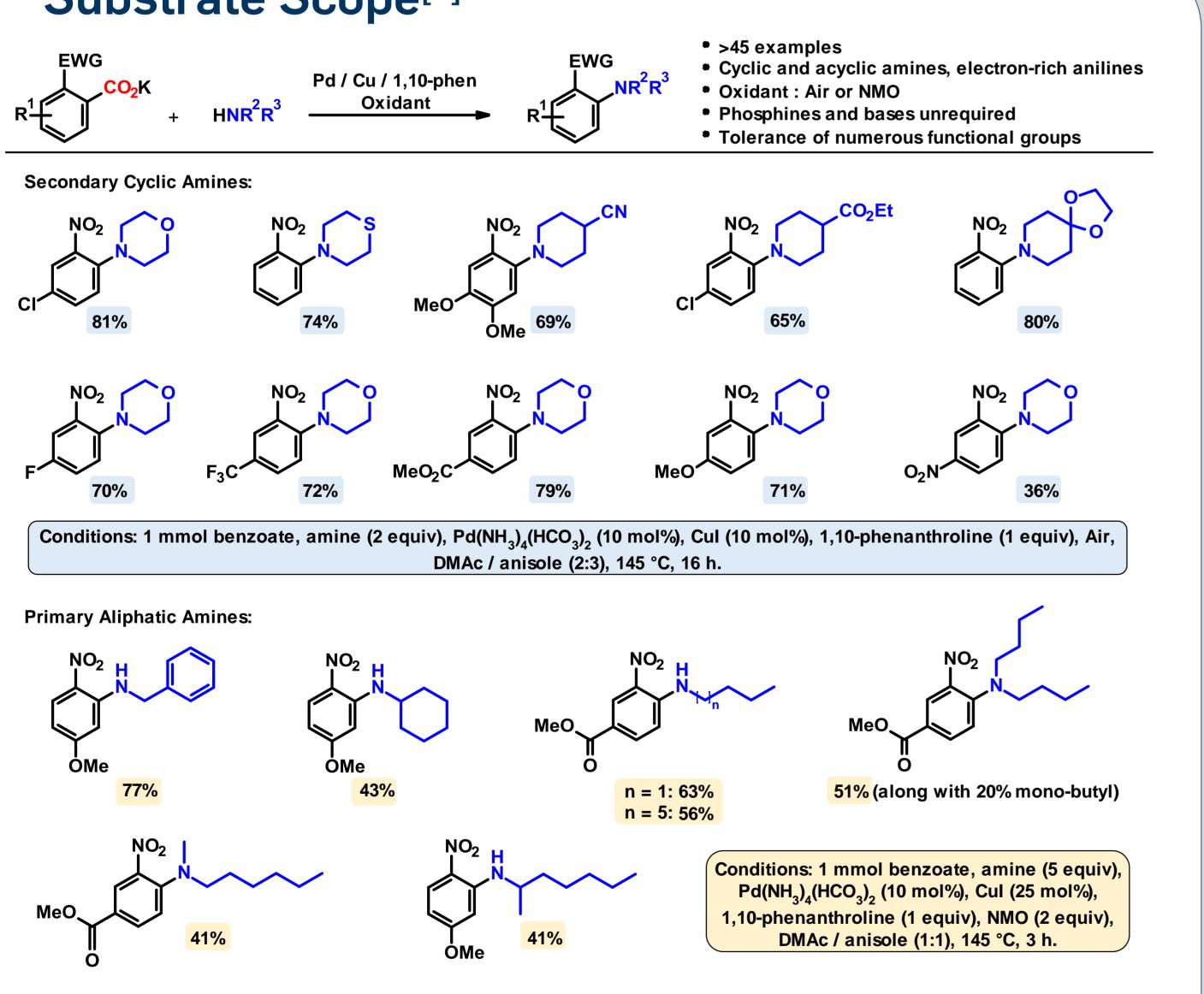


Proposed Mechanism $L_2Pd^{II}X_2$ amine $L_2Pd^{II}X_2$ binding $L_2Pd^{II}X_2$ > Pd/Cu bimetallic system _Ar[Cu] ArNR₂ reductive trans-ArCO₂[Cu] > 1, 10-phenanthroline as ligand for elimination metalation both metals СиХ L_2X_2Pd ArCO₂K \succ Oxidation via NMO or even O₂ from Air oxidation \geq Neither additional base nor phosphine ligands are required **Mechanistic Investigations** Amine binding



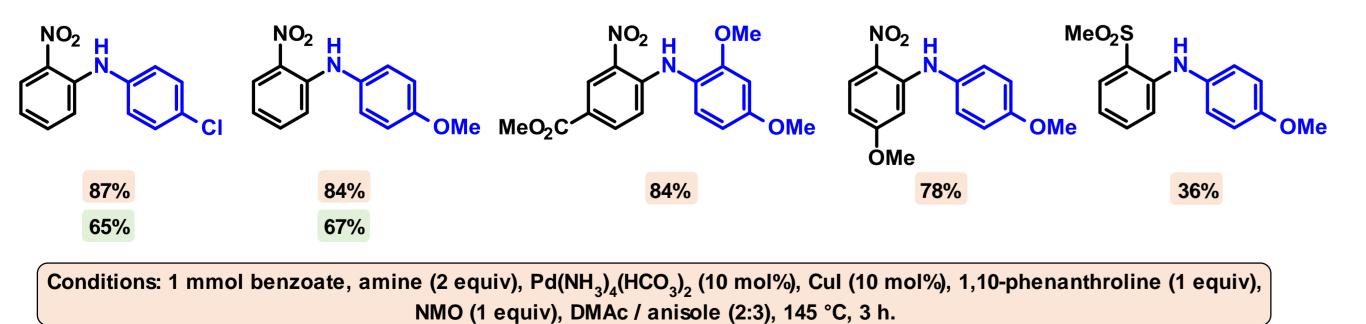


Substrate Scope^[7]

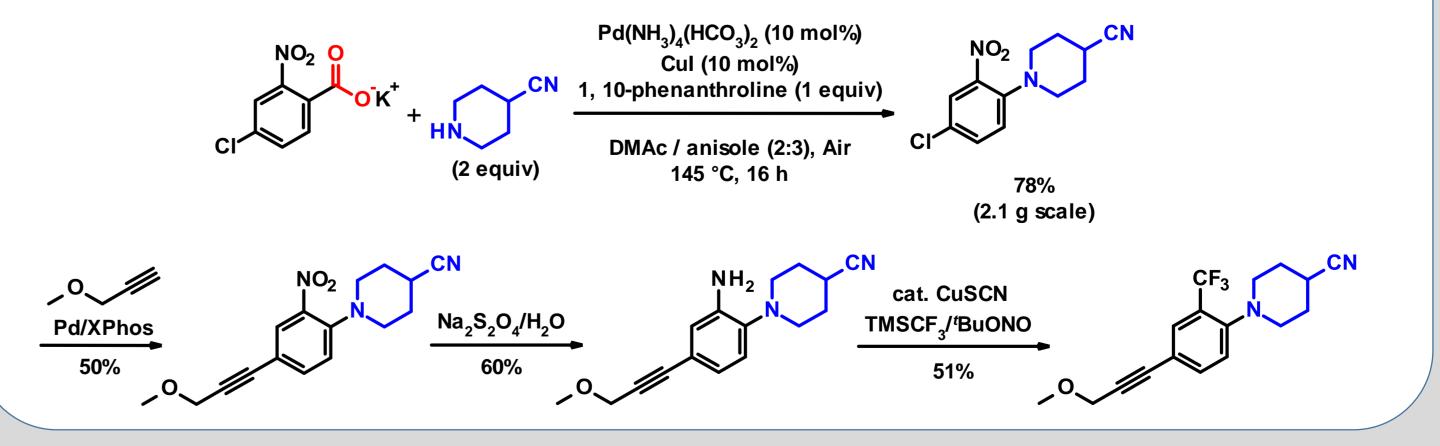


Orthogonal Catalytic Arylations

synthetic opportunities opened up by the orthogonality of the The decarboxylative *ipso*-amination to other Pd-catalyzed cross-coupling reactions are showcased by the following multistep synthesis. Multisubstituted arenes can be accessed via sequential C-C, C-Cl and C-N functionalizations.



Conditions: 1 mmol benzoate, amine (1.2 equiv), $Pd(NH_3)_4(HCO_3)_2$ (10 mol%), Cul (10 mol%), 1,10-phenanthroline (0.5 equiv), NMO (1 equiv), NMP, 120 °C, 16 h.



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